

Rearrangement of the intermediate 1,4-biradicals in photocycloaddition of cyclohex-2-enones to alkenes*

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Two unusual examples of enone/alkene photocycloaddition involving a rearrangement of the intermediate 1,4-biradical are presented. The first reaction proceeds *via* the addition of one of the radical centers to a carbonyl C atom and subsequent bond cleavage, *i.e.*, with rearrangement to a 1,3-biradical, while the second reaction involves abstraction of an H atom by one of the radical centers.

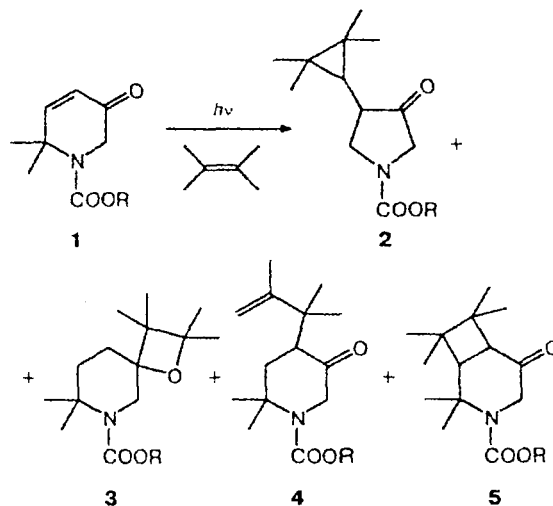
Key words: 1,4-biradicals, 1,3-biradicals, piperidine/pyrrolidine ring contraction, biradical/biradical rearrangement.

Triplet tetramethylene 1,4-biradicals are easily accessible upon the addition of an alkene (or an alkene fragment) to one of the C atoms of the C=C double bond in the triplet-excited cyclohex-2-enone.¹ The nonradiative intersystem crossing of the triplet biradical affords a singlet biradical, which either undergoes 1,4-cyclization to give cyclobutane or decomposes to give the starting compounds. The factors affecting the distribution between these two alternative reactions seem to be well understood.^{2–4} Here we report on two unusual previously unknown rearrangements of these intermediates.

Photoaddition of 2,2-dimethyl-5-oxo-1,2,5,6-tetrahydropyridine-1-carboxylates (1) to 2,3-dimethylbut-2-ene

Irradiation ($\lambda > 340$ nm) of ethyl carbamate **1a** in benzene in the presence of excess 2,3-dimethylbut-2-ene unexpectedly afforded cyclopropylpyrrolidine **2a** (50%) as the main product,⁵ in addition to oxetane **3a** (25%) and a mixture (1 : 1) of the expected⁶ photoaddition (**4a**) and photocycloaddition (**5a**) products (25%) (Scheme 1). Methyl carbamate **1b** reacts in a similar way, giving a ring contraction product, pyrrolidine **2b** (30%), together with oxetane **3b** (28%) and a mixture of compounds **4b** and **5b** (42%) (GLC data). The structures of the products were confirmed by GLC/MS analysis and ¹H NMR spectroscopy. Indeed, the cyclopropane ring proton in **2b** resonates at 0.16 ppm (d, $J = 10.7$ Hz), and the olefinic protons in oxetane **3b** are manifested at 6.62 and 6.00 ppm (both d, $J = 10.2$ Hz).

Scheme 1



R = Et (a), Me (b)

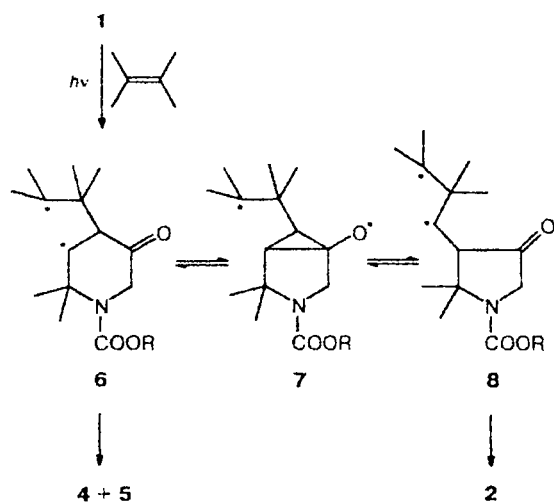
We suggested⁵ that the COOR group at the ring N atom increases the steric hindrance to 1,4-cyclization of biradical **6** due to the presence of two vicinal methyl groups. Therefore, **6** rearranges *via* biradical **7** to give biradical **8** and the latter cyclizes to cyclopropane **2** (Scheme 2). In our opinion, this mechanism is supported by the fact that the proportion of product **2** decreases when the ethoxycarbonyl group at the ring N atom is replaced by a smaller methoxycarbonyl group.

Photoisomerization of 3-(3,3-dimethylbut-1-ynyl)-4-(pent-4-enyl)cyclohex-2-enone (9)

Irradiation ($\lambda > 340$ nm) of cyclohex-2-enone **9** gives at low degrees of conversion (<40%) two isomeric satu-

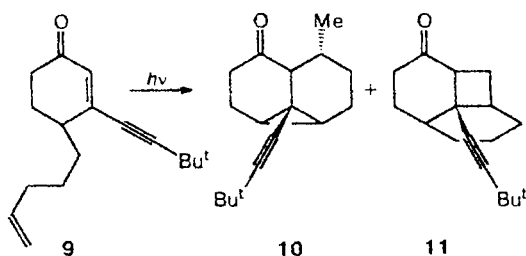
*The material was partly presented as an invited lecture by Paul Margaretha at the Sixth International Conference on the Chemistry of Carbenes and Related Intermediates (St. Petersburg, May 28–30, 1998).

Scheme 2



rated ketones **10** and **11**, containing intact $\text{C}\equiv\text{C}$ triple bonds,^{7,8} in 2.5 : 1 ratio (Scheme 3). On subsequent irradiation at the same wavelength, product **10** undergoes photodecomposition. The minor product **11** results from intramolecular [2+2]-cycloaddition, in which the terminal CH_2 group of the alkene fragment has added to the C(2) atom of the enone.⁹ In the ^1H and ^{13}C NMR spectra of compound **10** (in CD_3CN), the signals of the formed methyl group (δ 1.15, d, $J = 7$ Hz (^1H NMR) and δ 21.8 (^{13}C NMR)) immediately attract attention. The ^1H , ^{13}C COSY spectrum allows assignment of four CH and four CH_2 groups; however, the ^1H , ^1H COSY spectrum cannot be fully interpreted due to the partial overlap of the proton signals in the ^1H NMR spectrum. Nevertheless, it is evident that the CH group adjacent to the methyl group is bound to the CH group that is vicinal to the carbonyl C atom.

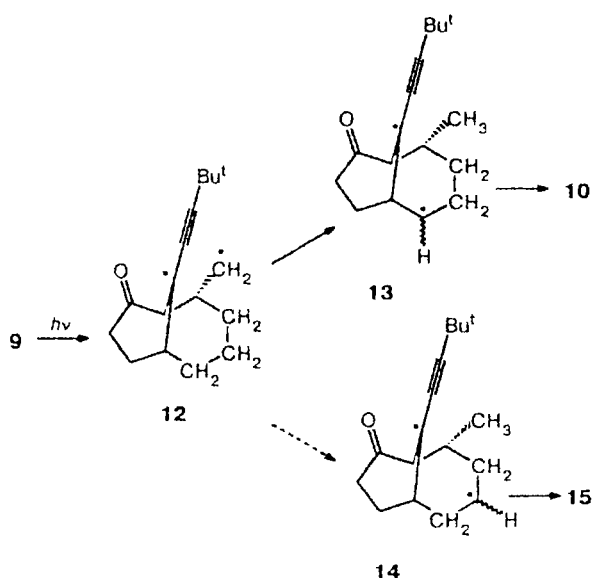
Scheme 3



The spectroscopic features mentioned above do not allow unambiguous assignment of the structure of **10**. However, it is evident that **10** is formed from biradical **12**, which isomerizes to a new biradical upon hydrogen abstraction by the primary radical center. It is well

known that in flexible chains, hydrogen abstraction *via* a six-membered transition state (formation of biradical **13**) occurs faster than that *via* a five-membered transition state (formation of biradical **14**); therefore, in our opinion, the pathway leading to a cyclopropane ring closure (product **10**) is preferred over the alternative pathway that would afford tricyclo[7.1.1.0^{5,9}]undecan-4-one (**15**) (Scheme 4).

Scheme 4



Experimental

Photolysis was performed in a Rayonet RPR-100 photo-reactor equipped with a 350-nm lamp using an additional light filter ($\lambda < 340$ nm). Analytical GLC was carried out on a 30-m long capillary column with SE-30 as the stationary phase. UV spectra were recorded on a Perkin Elmer Lambda 20 spectrophotometer. NMR spectra were run on a Bruker DRX-500 spectrometer (500 MHz for ^1H and 100.62 MHz for ^{13}C) in CDCl_3 ; chemical shifts were referred to internal tetramethylsilane. The positions of signals, whose multiplicity is not given, were derived from analysis of ^1H , ^1H and ^1H , ^{13}C COSY spectra.

Methyl 2,2-dimethyl-5-oxo-1,2,5,6-tetrahydropyridine-1-carboxylate (1b) was synthesized from methyl 2-[N-(1,1-dimethyl-2-oxopropyl)amino]ethanoate as reported previously⁵ for **1a** (four steps; overall yield 12%), m.p. 45 °C. UV (C_6H_{12}), λ/nm (log ϵ): 340 (1.821) and 242 (3.643). ^1H NMR, δ : 1.64 (s, 6 H, CH_3), 3.73 (s, 3 H, OCH_3), 4.10 (s, 2 H, CH_2), 6.63 and 6.01 (AB system, 2 H, $J = 10.7$ Hz). ^{13}C NMR (CDCl_3), δ : 24.0 (q, CH_3), 49.0 (t, CH_2), 52.0 (q, CH_3O), 55.0 (s, $\text{C}-\text{Me}_2$), 122.0 (s, $\alpha-\text{C}=\text{C}$), 156.0 (s, COO), 157.0 (s, $\beta-\text{C}=\text{C}$), 192.0 (s, CO).

3-(3,3-Dimethylbut-1-ynyl)-4-(pent-4-enyl)cyclohex-2-enone (9). 3-Methoxycyclohex-2-enone (12.6 g, 0.1 mol) was first converted into 3-methoxy-6-(pent-4-enyl)cyclohex-2-enone by low-temperature alkylation with 5-iodopent-1-ene

in the presence of LDA/HMPA. Then the product was treated with 3,3-dimethylbut-1-ynylmagnesium bromide followed by acidification and purification by chromatography (SiO_2 , hexane—EtOAc, 2 : 1) to give compound **9** (in 31% yield over 2 steps) as a colorless oil, $R_f = 0.70$. UV (C_6H_{12}), λ/nm (log ϵ): 340 (1.863). ^1H NMR, δ : 6.10 (d, 1 H, $\text{CH}=\text{}$, $J = 1.5$ Hz). ^{13}C NMR (CDCl_3), δ : 26.0 (t, CH_2), 27.0 (t, CH_2), 28.0 (s, CMe_3), 30.0 (q, CH_3), 32.0 (t, CH_2), 34.0 (t, CH_2), 35.0 (t, CH_2), 39.0 (d, ring C(4)), 78.0 (s, C(1)=), 111.0 (s, C(2)=), 115.0 (t, side-chain C(5)), 132.0 (d, side-chain C(4)), 138.0 (d, ring C(2)), 149.0 (s, ring C(3)), 199.0 (s, CO).

Irradiation of compound 1b. An argon-degassed solution of **1b** (18.3 mg, 0.1 mmol) and 2,3-dimethylbut-2-ene (168 mg, 2 mmol) in 5 mL of benzene was irradiated for 2 h. The products were analyzed by GLC, GLC/MS, and ^1H NMR.

Irradiation of compound 9. An argon-degassed solution of **9** (244 mg, 1 mmol) in 10 mL of benzene was irradiated for 1 h to give a mixture of compounds **9** (65%), **10** (25%), and **11** (10%). Chromatography (SiO_2 , hexane—EtOAc, 2:1) gave compound **9** and then the mixture of **10** and **11** (R_f 0.40). This second fraction was separated either by flash chromatography or by spinning-disk chromatography (hexane—EtOAc, 20 : 1, as the eluent). This gave 25 mg of 10-*exo*-(3,3-dimethylbut-1-ynyl)-6-*endo*-methyltricyclo[7.1.0.0^{5,10}]decan-4-one (**10**) as a colorless oil and 9 mg of 10-(3,3-dimethylbut-1-ynyl)tricyclo[7.1.1.0^{5,10}]undecan-2-one (**11**).

Compound **10**. ^1H NMR (CD_3CN), δ : 1.15 (d, 3 H, CH_3 , $J = 7$ Hz), 1.20 (s, 9 H, Bu^t), 1.64 (HCH), 1.75 (HCH), 1.89 (HCH), 1.93 (CH), 1.95 (HCH), 1.97 (HCH), 2.07 (HCH), 2.15 (1 H, HCH), 2.26 (1 H, CH), 2.31 (1 H, CH), 2.42 (1 H, CH), 2.56 (1 H, HCH). ^{13}C NMR (CD_3CN), δ : 21.0 (q, CH_3), 24.0 (t, CH_2), 27.0 (s, CMe_3), 29.0 (t, CH_2), 30.0

(t, CH_2), 31.0 (q, CH_3), 34.0 (t, CH_2), 35.0 (d, CH), 42.0 (s, $\text{CC}=\text{}$), 43.0 (d, CH), 54.0 (d, CH), 56.0 (d, CH), 85.0 (s, C(1)=), 90.0 (s, C(2)=), 211.0 (s, CO).

Compound **11**. ^1H NMR (C_6D_6), δ : 1.19 (s, 9 H, Bu^t), 1.35–1.20 (m, 5 H), 1.46 (m, 1 H), 1.57 (m, 1 H), 1.93 (2 H), 2.05 (1 H, HCH), 2.37 (1 H, HCH), 2.75 (1 H, HCH), 3.06 (dd, H(1), $J = 9, 10$ Hz). ^{13}C NMR (C_6D_6), δ : 15.0 (t, CH_2), 24.0 (t, CH_2), 25.0 (t, CH_2), 25.5 (t, CH_2), 26.0 (s, CMe_3), 27.0 (t, CH_2), 30.0 (q, CH_3), 35.0 (s, $\text{CC}=\text{}$), 36.0 (d, CH), 37.0 (t, CH_2), 38.0 (d, CH), 48.0 (d, CH), 85.0 (s, C(1)=), 88.0 (s, C(2)=), 207.0 (s, CO).

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